

25.3

10. Melting point (°C) of aluminum 660, 667, 654, 663, 662

 $N=5$

$$E[x] = \frac{660 + 667 + 654 + 663 + 662}{5} = 661.2 = \mu$$

$$\begin{aligned} \text{Var}[x] &= E[(x - E[x])^2] \\ &= \frac{(-1.2)^2 + (5.8)^2 + (-7.2)^2 + (1.8)^2 + (0.3)^2}{5} \end{aligned}$$

$$= \frac{1.44 + 33.64 + 51.84 + 3.24 + 0.09}{5}$$

$$= \frac{90.3}{5} = 18.06$$

$$\sigma = \sqrt{\text{Var}[x]} = \sqrt{18.06} = 4.26$$

$$\sigma_n = \frac{\sigma}{\sqrt{N}} = \frac{4.26}{\sqrt{5}} = 1.9$$

2.326 for 99% confidence interval

$$\mu \pm 2.326 \sigma_n$$

$$661.2 \pm 4.43 = [656.77, 665.63]$$

if we remove 654

$$E[x] = \mu = \frac{660 + 667 + 662 + 663}{4} = 663$$

 $N=4$

$$\text{Var}[x] = \frac{3^2 + 4^2 + 1^2 + 0}{4} = \frac{9 + 16 + 1}{4} = \frac{26}{4} = 6.5$$

$$\sigma = \sqrt{\text{Var}[x]} = \sqrt{6.5} = 2.55$$

$$\sigma_n = \frac{\sigma}{\sqrt{N}} = \frac{2.55}{2} = 1.275$$

$$\mu \pm 2.326 \sigma_n$$

$$663 \pm 2.326(1.275) = 663 \pm 2.97 = [660, 666]$$